

# ADMISSION CUM SCHOLARSHIP TEST SAMPLE TEST PAPER

(For Students Going to Class 11<sup>™</sup> IN 2024)

**STREAM: ENGINEERING | COURSE OFFERED: GROUND ZERO** 

Time: 2 hours Maximum Marks: 240

### INSTRUCTIONS

#### (A) General:

- 1. This Question paper contains **FOUR** Parts (Physics, Chemistry, Mathematics & Mental Ability) containing 60 questions in all.
- 2. This Question Paper contains 12 pages, other than the OMR.
- 3. This Question Paper contains total **60 questions**, **15 questions each** in **Physics**, **Chemistry**, **Mathematics & Mental Ability**.
- 4. The Question Paper has blank spaces at the bottom of each page for rough work. No additional sheets will be provided for rough work.
- 5. Blank papers, clip boards, log tables, slide rule, calculators, cellular phones, pagers and electronic gadgets, in any form, are **NOT** allowed.
- 6. This booklet also contains the **OMR** answer sheet (i.e., A machine gradable Response Sheet).

#### (B) Answering on the OMR:

- 7. Each question will have **4 choices** in both the Sections, out of which **only one choice is correct**.
- 8. Darken the bubble with **Ball Pen (Blue or Black) ONLY.**
- (C) Filling in Name and Registration No.
- 8. On the **OMR sheet**, write your Name and Registration No. in ink. Also, put your signature in the appropriate box in ink.

### (D) Marking Scheme:

DO NOT BREAK THE SEALS ON THIS BOOKLET, AWAIT INSTRUCTIONS FROM THE INVIGILATOR.

9.

- (a) For each question, you will be awarded 4 marks if you have darkened only one bubble corresponding to the right answer.
  - (b) In case you have not darkened any bubble, you will be awarded 0 mark for that question.
  - (c) In all other cases, you will be awarded -1 mark.

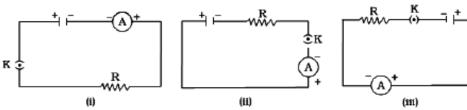
Name :	 			 	 	
Registration No.:	JLJL	_  _	_ L_ L	JLJL	JLLL	

(C) Secondary rainbow is brighter than primary rainbow.

(D) In secondary rainbow, light wave suffers one total internal reflection before coming out.



**8.** A cell, a resistor, a key and ammeter are arranged as shown in the circuit diagrams of Figure. The current recorded in the ammeter will be

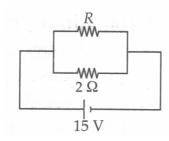


(A) maximum in (i)

(B) maximum in (ii)

(C) maximum in (iii)

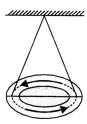
- (D) the same in all the cases
- **9.** What is the maximum resistance which can be made using five resistors each of  $1/5 \Omega$ ?
  - (A)  $1/5\Omega$
- (B)  $10\Omega$
- (C)  $5\Omega$
- (D)  $1\Omega$
- 10. If in the circuit, power dissipation is 150 W, then R is



- (A) 2  $\Omega$
- (B) 6  $\Omega$
- (C) 5  $\Omega$
- (D) 4 Ω

- 11. 1 horse power is equal to
  - (A) 700 W
- (B) 726 W
- (C) 736 W
- (D) 746 W
- **12.** If n equal resistances are first connected in series and then connected in parallel, the ratio of the maximum to the minimum resistance is
  - (A) n
- (B)  $\frac{1}{n^2}$
- (C) n<sup>2</sup>
- (D)  $\frac{1}{n}$

**13.** A circular loop is suspended in air as shown in figure. When the loop is seen from above, current flows anti-clockwise and when seen from below, current flows clockwise. This loop behaves as a magnet. The N-pole of this magnet is on



(A) the upper face

- (B) lower face
- (C) the lower face if current is large
- (D) upper face if current is large

- 14. SI unit of resistivity is
  - (A) ohm-meter
- (B) ohm-meter<sup>2</sup>
- (C) ohm<sup>-1</sup>
- (D) ohm-meter-1

- **15.** Which of the following is a better nuclear fuel?
  - (A) Thorium 236

(B) Uranium - 235

(C) Neptunium - 239

(D) Plutonium - 239

## **PART-B: CHEMISTRY**

- **16.** 10<sup>-6</sup> M HCl is diluted to 100 times. Its pH is
  - (A) 6.0
- (B) 8.0
- (C) 6.95
- (D) 9.5
- 17. Iron filings were added to solution of copper sulphate. After 10 minutes, it was observed that the blue colour of the solution changed and layer got deposited on iron filings. The colour of the solution and that of the layer would respectively be
  - (A) Yellow and green

(B) Brown and blue

(C) Red and greenish blue

(D) Green and reddish brown

- **18.** Write the net ionic equation for the reaction of sodium hydroxide with hydrochloric acid.
  - (A) Na<sup>+</sup> + Cl<sup>-</sup> → NaCl

(B) Na<sup>+</sup> + Cl<sup>-</sup> H<sup>+</sup> + OH<sup>-</sup>  $\rightarrow$  NaCl + H<sub>2</sub>O<sub>(I)</sub>

 $(C) H^+ + OH^- \rightarrow H_2O_{(1)}$ 

- (D) None of these
- 19. The hydrophilic part of a synthetic detergent is
  - (A) CH<sub>3</sub>(CH<sub>2</sub>)<sub>10</sub>-CH<sub>2</sub>-

(B) -CO-Na+

 $(C) -SO_3^-Na^+$ 

- (D) -COO-Na+
- 20. In the given structure, the type of carbon atoms present are

$$\begin{array}{c} & \text{CH}_3 \\ \mid \\ \text{H}_3\text{C} - \text{CH} - \text{CH}_2 - \text{C} - \text{CH}_3 \\ \mid \\ \text{CH}_3 & \text{CH}_3 \end{array}$$

- (A) One primary, two secondary and one tertiary
- (B) Four primary, one secondary and two tertiary
- (C) One primary, one secondary, one tertiary and one quaternary
- (D) Five primary, one secondary, one tertiary and one quaternary
- 21. The IUPAC name of

$$\begin{array}{c} H \\ | \\ CH_3 - C - CH_2 - CH_2 - CH_2 - C - CH_3 \\ | \\ OH \end{array} \quad \text{is} \quad$$

- (A) 6, 6 -dibromoheptan-2-ol
- (B) 2, 2-dibromoheptan-2-ol

(C) 6, 6-dibromoheptan-2-al

- (D) 2, 6-dibromohydroxy heptane
- 22. Which of the following statements is not correct?
  - (A) All metals are solid at room temperature.
  - (B) All metals are good conductors of heat and electricity.
  - (C) All metals form basic oxides.
  - (D) All metals possess lustre when freshly prepared.

[6]					SAMPLE PAPER (Ground Ze
23.	The metal that re	acts with cold water is			
	(A) Mercury	(B) Sodium	(C) Zind	С	(D) Tungsten
24.	Brass is a mixture	e of			
	(A) Copper and z	inc			
	(B) Copper and til	n			
	(C) Copper, nicke	l and zinc			
	(D) Aluminium, co	opper and traces of Mg	and Mn		
25.	Among Al <sub>2</sub> O <sub>3</sub> , SiO	O <sub>2</sub> , P <sub>2</sub> O <sub>3</sub> and SO <sub>2</sub> the co	rrect order of	acid stren	gth is
	(A) $SO_2 < P_2O_3 <$	$SiO_2 < Al_2O_3$	(B) Al <sub>2</sub> (	O <sub>3</sub> < SiO <sub>2</sub> <	$P_2O_3 < SO_2$
	(C) Al2O3 < SiO2	< SO2 < P2O3	(D) SiC	o <sub>2</sub> < SO <sub>2</sub> < A	$AI_2O_3 < P_2O_3$
26.	The correct order	of electron affinity amo	ng the followi	ng is	
	(A) F > Cl > Br	(B) Br > Cl > F	(C) Cl >	> F > Br	(D) F > Br > Cl
27.	Alkali metals in ea	ich period have			
	(A) Smallest size		(B) Lov	vest I.E.	
	(C) Highest I.E.		(D) Hig	hest electro	onegativity
28.	In the balanced ed	quation			
	$Cu + xHNO_3 \rightarrow C$	$u(NO_3)_2 + yNO_2 + 2H_2O_3$	)		
	The values of x ar	nd v are			
	(A) 3 and 5	(B) 8 and 6	(C) 4 aı	nd 2	(D) 7 and 1
29.	,	• •	, ,		elop a green layer which
	(A) basic copper	carbonate	(B) cop	per sulpha	te
	(C) copper carbor	nate	(D) cop	per nitrate	
		C	or rough work		

- **30.** In the reaction  $PCl_3 + Cl_2 \rightarrow PCl_5$ 
  - (A) PCl<sub>3</sub> is acting as reductant
  - (B) Cl<sub>2</sub> is acting as reductant
  - (C) both PCl<sub>3</sub> and Cl<sub>2</sub> are acting as reductant
  - (D) both PCl<sub>3</sub> and Cl<sub>2</sub> are acting as oxidant.

## PART-C: MATHEMATICS

- 31. The HCF of 24, 32, and 60 is equal to
  - (A) 4

- (C) 8
- (D) 3

- **32.** The factors of  $\left(\frac{1}{3}x^2 2x 9\right)$  is equal to

- (A)  $\frac{1}{3}(x-9)(x+3)$  (B)  $\frac{1}{3}(x-9)(x-3)$  (C)  $\frac{1}{3}(x+9)(x+3)$  (D)  $\frac{1}{3}(x+9)(x-3)$
- **33.** The solution of the system of equations x + 3y = 4 and 4x + 7y = 1 is
  - (A) x = -5 and y = 3

(B) x = -2 and y = 7

(C) x = 1 and y = 1

- (D) x = 5 and y = 4
- **34.** If (3 + i) is a root of the equation  $x^2 + ax + b = 0$  (where a, b  $\in \mathbb{R}$ ) then a is equal to
- (B) -3
- (C) 6

- (D) -6
- **35.** The fourth term of an A.P. is 4. Then the sum of the first 7 terms is
  - (A)4
- (B) 28

- (C) 16
- (D) 40
- **36.** If  $\tan \theta = \frac{1}{\sqrt{7}}$  and  $\theta$  is an acute angle, then  $\frac{\csc^2 \theta \sec^2 \theta}{\csc^2 \theta + \sec^2 \theta} \alpha$  is equal to
  - (A)  $\frac{3}{4}$
- (B)  $\frac{1}{2}$

(C)2

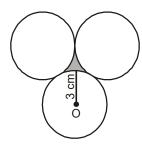
(D)  $\frac{5}{4}$ 

- 37. The distance of the point P(2, 3) from the x-axis is
  - (A) 2

(B) 3

(C) <sup>2</sup>

- (D) 5
- **38.** If  $\sin\theta \cos\theta = 0$ , then the value of  $(\sin^4\theta + \cos^4\theta)$  is
  - (A) 1
- (B)  $\frac{3}{4}$
- (C)  $\frac{1}{2}$
- (D)  $\frac{1}{4}$
- **39.** If three equal circles of radius 3 cm each touch each other externally as shown, then the area of the shaded portion is :



(A)  $\frac{\sqrt{3}}{2}(2-\pi) \text{ cm}^2$ 

(B)  $\frac{9}{2}(2\sqrt{3}-\pi) \text{ cm}^2$ 

(C)  $\frac{9}{2}(2\sqrt{3} + \pi)$  cm<sup>2</sup>

- (D)  $\frac{3}{2}(\sqrt{3}-\pi) \text{ cm}^2$
- **40.** The number of observations in a group is 40. If the average of first 10 is 4.5 and that of the remaining 30 is 3.5, then the average of the whole group is equal to
  - (A)  $\frac{1}{5}$
- (B)  $\frac{15}{4}$

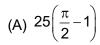
(C) 4

- (D) 8
- 41. An unbiased die is thrown, then the probability of getting a number greater than 1 is
  - (A)  $\frac{1}{6}$
- (B)  $\frac{2}{6}$

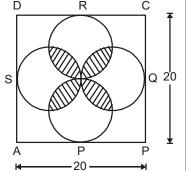
- (C)  $\frac{4}{6}$
- (D)  $\frac{5}{6}$

- The length of the parallel sides of a trapezium are 14 cm and 7 cm. If the length of third side is 42. 8 cm and of fourth side is x cm, then the number of possible intergral value of x is:
  - (A) 12
- (B) 13

- (C) 14
- (D) 17
- In the fig. below, ABCD is a square and 4 congruent circles are inscribed inside it such that each of the circles is touching the sides of the square at its mid-point. Then evaluate the area of the shaded region (in sq. units):



- (B)  $50(\pi-1)$
- (C)  $100\left(\frac{\pi}{2}-1\right)$
- (D)  $200(\pi-1)$



- **44.** If  $\sin \theta$  and  $\cos \theta$  are the roots of the equation  $ax^2 bx + c = 0$ , then a, b and c satisfy the relation
  - (A)  $a^2 + b^2 + 2ac = 0$

(B)  $a^2 - b^2 + 2ac = 0$ 

 $(C) a^2 + c^2 + 2ab = 0$ 

- (D)  $a^2 b^2 2ac = 0$
- If  $\sin \theta = \sin \alpha$  then the value of  $\sin \frac{\theta}{3}$  can be 45.
- (A)  $\sin \frac{\alpha}{3}$  (B)  $\cos \left(\frac{\pi}{3} \frac{\alpha}{3}\right)$  (C)  $\csc \left(\frac{\pi}{3} + \frac{\alpha}{3}\right)$  (D) none of these

## **PART-D: MENTAL ABILITY**

## Find the missing number:

46.

78	?	97
43	67	58
35	13	39

(A) 84

(B) 80

(C) 54

(D) 48

**47**. <sup>3</sup> 6 6 5



3 ? 7

(A) 4

(B) 9

(C) 6

(D) 8

**48**. <sup>4</sup>



10



(A) 9

(B) 8

(C) 6

(D) 5

- **49.** If in a code language **MENTAL** is coded as 417253, then how is **TEN & ANT** coded in that language?
  - (A) 572,271
- (B) 217,527
- (C) 572, 217
- (D) 217, 572
- **50.** If **REASON** is coded as 5 and **BELIEVED** as 7, what is the code number for **GOVERNMENT**?
  - (A) 6
- (B) 8

(C) 9

(D) 10

- 51. If E = 5 & SAFE = 31, then PINK = ?
  - (A) 41
- (B) 40

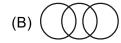
- (C) 50
- (D) 65

SAMP	LE PAPER (Ground Zero)				[ 11 ]
52.	walks 30 metre. The	n moves 25 metre to his	righ	t. He then turns to	then turns to his left and his right again and walks direction is he now from
	(A) South-West	(B) South	(C)	North-West	(D) South-East
53.					owards South. Then, he reference to his starting
	(A) 5 km, North	(B) 5 km, North-East	(C)	7 km, East	(D) 7 km, West
54.	•	•		•	ralked another two paces nich direction was he last
	(A) East	(B) North	(C)	South	(D) None of these
55.		graph, a man says to his ". How is the girl in the pl		•	nd-daughter of the elder e man ?
	(A) Niece	(B) Sister	(C)	Aunt	(D) Sister-in-law
<b>56</b> .		the photograph. Amar sa w is the girl's mother rel			ner is the only son of my
	(A) Mother	(B) Sister	(C)	Sister-in-law	(D) Grandmother
57.	T is the son of P.S. to T?	is the son of Q. T is ma	rried	to R. R is Q's da	ughter. How is S related
	(A) Brother	(B) Uncle	(C)	Father-in-law	(D) Brother-in-law
		Space for ro	ugh w	ork	

DIRECTIONS: (58 to 60) Each of these questions given below contains three group of things. You are to choose from the following four numbered diagrams, a diagram that depicts the correct relationship among the three groups of thing in each question.











59. India, Pakistan, Asia



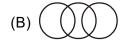






60. Batsman, Cricket, Stick









# ANSWER KEY SAMPLE TEST PAPER

(For Students Going to Class 11<sup>™</sup> IN 2024)

**STREAM**: ENGINEERING | COURSE OFFERED: GROUND ZERO

		F	PHYSICS	
1.	(A)	2. (A)	3. (D)	4. (B)
5.	(C)	6. (C)	7. (A)	8. (D)
9.	(D)	10. (B)	11. (D)	12. (C)
13.	(A)	14. (A)	15. (B)	
		CH	HEMISTRY	
16.	(C)	17. (D)	18. (D)	19. (C)
20.	(D)	21. (A)	22. (A)	23. (B)
24.	(A)	25. (B)	26. (C)	27. (B)
28.	(C)	29. (A)	30. (A)	
		MA	THEMATICS	
31.	(A)	32. (A)	33. (A)	34. (D)
35.	(B)	36. (A)	37. (B)	38. (C)
39.	(B)	40. (B)	41. (D)	42. (B)
43.	(C)	44. (B)	45. (A)	
		MEN <sup>-</sup>	TAL ABILITY	
46.	(B)	47. (C)	48. (D)	49. (D)
<b>50</b> .	(C)	51. (C)	52. (D)	53. (B)
54.	(A)	55. (A)	56. (A)	57. (D)
<b>5</b> 8.	(A)	59. (A)	60. (D)	