

## ADMISSION CUM SCHOLARSHIP TEST SAMPLE TEST PAPER

(For Students Going to Class 11<sup>™</sup> IN 2024)

**STREAM**: MEDICAL | COURSE OFFERED: MEDICAL GROUND ZERO

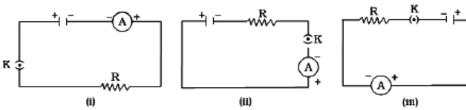
Time: 2 hours **Maximum Marks: 240** 

### INSTRUCTIONS DO NOT BREAK THE SEALS ON THIS BOOKLET, AWAIT INSTRUCTIONS FROM THE INVIGILATOR. (A) General: 1. This Question paper contains FOUR Parts (Physics, Chemistry, Biology & Mental Ability) containing 60 questions in all. This Question Paper contains 12 pages, other than the OMR. 3. This Question Paper contains total 60 questions, 15 questions each in Physics, Chemistry, Biology & Mental Ability. The Question Paper has blank spaces at the bottom of each page for rough work. No additional sheets will be provided for rough work. 5. Blank papers, clip boards, log tables, slide rule, calculators, cellular phones, pagers and electronic gadgets, in any form, are **NOT** allowed. This booklet also contains the **OMR** answer sheet (i.e., A machine gradable Response Sheet). 6. (B) Answering on the OMR: 7. Each question will have 4 choices in both the Sections, out of which only one choice is correct. 8. Fill the bubble with Ball Pen (Blue or Black) ONLY. (C) Filling - Name and Registration No. 9. On the **OMR sheet**, write your Name and Registration No. using ball pen. Also, put your signature in the appropriate box using ball pen. (D) Marking Scheme: (a) For each question, you will be awarded 4 marks if you have darkened only one bubble corresponding to the right answer. (b) In case you have not darkened any bubble, you will be awarded 0 mark for that question. (c) In all other cases, you will be awarded **-1 mark**.

**Registration No.:** 

		PART-A:	PHYSICS	
1.		ages formed by two pla lly between mirrors is	ine mirrors inclined at	an angle 60° of an object
	(A) 5	(B) infinite	(C) 6	(D) 7
2.	A convex mirror has it from the pole, pro	•	A real object placed at	a distance 15 cm in front of
	(A) 7.5 cm	(B) 30 cm	(C) infinity	(D) 15 cm
3.	<u> </u>	the size is obtained in a noce of the object and the	•	or with a radius of curvature.
	(A) 15 cm, 30 cm	(B) 30 cm, 15 cm	(C) 30 cm, 60 cm	(D) 60 cm, 30 cm
4.	-	reflected first from one	•	h other. A ray of light travel- the other mirror. Then the
	(A) 180°	(B) 240°	(C) 60°	(D) 120°
5.		•	9	an object are formed on the ratio between the sizes of
	(A) D/d		(B) D <sup>2</sup> /d <sup>2</sup>	
	$(C) (D - d)^2/(D + d)^2$	2	(D) $\sqrt{(D/d)}$	
6.	The property of per	sistence of vision is use	ed in	
	(A) short sightedne	ss	(B) long sightedne	SS
	(C) cinematography	/	(D) colour vision	
7.	Which of the follow	ing statements is correc	ct about rainbow ?	
	(A) In primary rainb	ow, red colour is on the	outside and violet cold	our is on the inside.
	(B) In primary rainb	ow, violet colour is on th	ne outside and red colo	our is on the inside.
	(C) Secondary rain	bow is brighter than prin	nary rainbow.	
	(D) In secondary ra	inbow, light wave suffer	s one total internal refle	ection before coming out.
		Space for	rough work	

**8.** A cell, a resistor, a key and ammeter are arranged as shown in the circuit diagrams of Figure. The current recorded in the ammeter will be

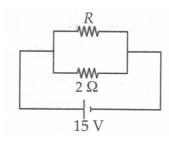


(A) maximum in (i)

(B) maximum in (ii)

(C) maximum in (iii)

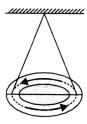
- (D) the same in all the cases
- **9.** What is the maximum resistance which can be made using five resistors each of  $1/5 \Omega$ ?
  - (A)  $1/5\Omega$
- (B)  $10\Omega$
- (C)  $5\Omega$
- (D)  $1\Omega$
- 10. If in the circuit, power dissipation is 150 W, then R is



- (A) 2  $\Omega$
- (B) 6  $\Omega$
- (C) 5  $\Omega$
- (D) 4 Ω

- 11. 1 horse power is equal to
  - (A) 700 W
- (B) 726 W
- (C) 736 W
- (D) 746 W
- **12.** If n equal resistances are first connected in series and then connected in parallel, the ratio of the maximum to the minimum resistance is
  - (A) n
- (B)  $\frac{1}{n^2}$
- (C) n<sup>2</sup>
- (D)  $\frac{1}{n}$

**13.** A circular loop is suspended in air as shown in figure. When the loop is seen from above, current flows anti-clockwise and when seen from below, current flows clockwise. This loop behaves as a magnet. The N-pole of this magnet is on



(A) the upper face

- (B) lower face
- (C) the lower face if current is large
- (D) upper face if current is large

- 14. SI unit of resistivity is
  - (A) ohm-meter
- (B) ohm-meter<sup>2</sup>
- (C) ohm<sup>-1</sup>
- (D) ohm-meter-1

- **15.** Which of the following is a better nuclear fuel?
  - (A) Thorium 236

(B) Uranium - 235

(C) Neptunium - 239

(D) Plutonium - 239

## **PART-B: CHEMISTRY**

- **16.** 10<sup>-6</sup> M HCl is diluted to 100 times. Its pH is
  - (A) 6.0
- (B) 8.0
- (C) 6.95
- (D) 9.5
- 17. Iron filings were added to solution of copper sulphate. After 10 minutes, it was observed that the blue colour of the solution changed and layer got deposited on iron filings. The colour of the solution and that of the layer would respectively be
  - (A) Yellow and green

(B) Brown and blue

(C) Red and greenish blue

(D) Green and reddish brown

- **18.** Write the net ionic equation for the reaction of sodium hydroxide with hydrochloric acid.
  - (A) Na<sup>+</sup> + Cl<sup>-</sup> → NaCl

(B) Na<sup>+</sup> + Cl<sup>-</sup> H<sup>+</sup> + OH<sup>-</sup>  $\rightarrow$  NaCl + H<sub>2</sub>O<sub>(1)</sub>

 $(C) H^+ + OH^- \rightarrow H_2O_{(1)}$ 

- (D) None of these
- 19. The hydrophilic part of a synthetic detergent is
  - (A) CH<sub>3</sub>(CH<sub>2</sub>)<sub>10</sub>-CH<sub>2</sub>-

(B) -CO-Na+

 $(C) -SO_3^-Na^+$ 

- (D) -COO-Na+
- 20. In the given structure, the type of carbon atoms present are

$$\begin{array}{c} & \text{CH}_3 \\ \mid & \mid \\ \text{H}_3\text{C} - \text{CH} - \text{CH}_2 - \text{C} - \text{CH}_3 \\ \mid & \mid \\ \text{CH}_3 & \text{CH}_3 \end{array}$$

- (A) One primary, two secondary and one tertiary
- (B) Four primary, one secondary and two tertiary
- (C) One primary, one secondary, one tertiary and one quaternary
- (D) Five primary, one secondary, one tertiary and one quaternary
- 21. The IUPAC name of

$$\begin{array}{c} H \\ | \\ CH_3 - C - CH_2 - CH_2 - CH_2 - C - CH_3 \\ | \\ OH \end{array} \quad \text{is} \quad$$

- (A) 6, 6 -dibromoheptan-2-ol
- (B) 2, 2-dibromoheptan-2-ol

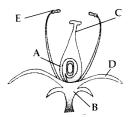
(C) 6, 6-dibromoheptan-2-al

- (D) 2, 6-dibromohydroxy heptane
- 22. Which of the following statements is not correct?
  - (A) All metals are solid at room temperature.
  - (B) All metals are good conductors of heat and electricity.
  - (C) All metals form basic oxides.
  - (D) All metals possess lustre when freshly prepared.

The metal that react (A) Mercury Brass is a mixture of (A) Copper and zinct (B) Copper and tin (C) Copper, nickel at (D) Aluminium, copper	(B) Sodium  of  on  and zinc	(C) Zinc	SAMIFLE	E PAPER (Medical Ground Zero  (D) Tungsten
<ul><li>(A) Mercury</li><li>Brass is a mixture of</li><li>(A) Copper and zince</li><li>(B) Copper and tine</li><li>(C) Copper, nickel at</li><li>(D) Aluminium, copper</li></ul>	(B) Sodium  of  on  and zinc	(C) Zinc		(D) Tungsten
Brass is a mixture of (A) Copper and zind (B) Copper and tind (C) Copper, nickel ad (D) Aluminium, copper	nd zinc	(C) Zinc		(D) Tungsten
(A) Copper and zind (B) Copper and tin (C) Copper, nickel a (D) Aluminium, copp	nd zinc			
(B) Copper and tin (C) Copper, nickel a (D) Aluminium, copp	nd zinc			
(C) Copper, nickel a				
(D) Aluminium, copp				
. ,	per and traces of Mg a			
Among ALO SiO	_			
	- ·			
2 2 0	2 2 0		,	2 0 2
(C) Al2O3 < SiO2 < S	$SO_2 < P_2O_3$	(D) SiO <sub>2</sub>	< SO <sub>2</sub> < Al	$_{2}O_{3} < P_{2}O_{3}$
The correct order of	electron affinity amor	ng the followin	g is	
(A) F > Cl > Br	(B) Br > Cl > F	(C) CI >	F > Br	(D) F > Br > Cl
Alkali metals in each	n period have			
(A) Smallest size		(B) Lowe	est I.E.	
(C) Highest I.E.		(D) High	est electron	egativity
In the balanced equa	ation			
$Cu + xHNO_3 \rightarrow Cu(I$	$(NO_3)_2 + yNO_2 + 2H_2O_3$			
The values of x and	y are			
(A) 3 and 5	(B) 8 and 6	(C) 4 and	12	(D) 7 and 1
Copper on exposure chemically	e to air reacts with me	oisture and Co	O <sub>2</sub> to devel	op a green layer which is
(A) basic copper carbonate		(B) coppersulphate		
(C) copper carbonat	te	(D) copp	er nitrate	
	Space fo	r rough work		
	(A) $SO_2 < P_2O_3 < SiO_2 < $	(A) $SO_2 < P_2O_3 < SiO_2 < Al_2O_3$ (C) $Al_2O_3 < SiO_2 < SO_2 < P_2O_3$ The correct order of electron affinity amore (A) $F > Cl > Br$ (B) $Br > Cl > F$ Alkali metals in each period have (A) Smallest size (C) Highest I.E. In the balanced equation $Cu + xHNO_3 \rightarrow Cu(NO_3)_2 + yNO_2 + 2H_2O$ The values of x and y are (A) 3 and 5 (B) 8 and 6 Copper on exposure to air reacts with machemically (A) basic copper carbonate (C) copper carbonate	(A) $SO_2 < P_2O_3 < SiO_2 < Al_2O_3$ (B) $Al_2O_3$ (C) $Al_2O_3 < SiO_2 < SO_2 < P_2O_3$ (D) $SiO_2$ The correct order of electron affinity among the following (A) $F > Cl > Br$ (B) $Br > Cl > F$ (C) $Cl > R$ Alkali metals in each period have (A) Smallest size (B) Lower (C) Highest I.E. (D) Highest I.E. (D) Highest In the balanced equation $Cu + xHNO_3 \rightarrow Cu(NO_3)_2 + yNO_2 + 2H_2O$ The values of x and y are (A) 3 and 5 (B) 8 and 6 (C) 4 and 6 (C) 4 and 6 (C) 4 and 6 (C) 4 basic copper carbonate (B) copp	$ (C) \operatorname{Al_2O_3} < \operatorname{SiO_2} < \operatorname{SO_2} < \operatorname{P_2O_3} \qquad \qquad (D) \operatorname{SiO_2} < \operatorname{SO_2} < \operatorname{Al_2O_3} $ The correct order of electron affinity among the following is $ (A) F > CI > Br \qquad (B) Br > CI > F \qquad (C) CI > F > Br $ Alkali metals in each period have $ (A) \operatorname{Smallest size} \qquad (B) \operatorname{Lowest I.E.} $ $ (C) \operatorname{Highest I.E.} \qquad (D) \operatorname{Highest electron} $ In the balanced equation $ \operatorname{Cu} + x\operatorname{HNO_3} \rightarrow \operatorname{Cu(NO_3)_2} + y\operatorname{NO_2} + 2\operatorname{H_2O} $ The values of x and y are $ (A) \operatorname{3 and 5} \qquad (B) \operatorname{8 and 6} \qquad (C) \operatorname{4 and 2} $ Copper on exposure to air reacts with moisture and $\operatorname{CO_2}$ to developmentally $ (A) \operatorname{basic copper carbonate} \qquad (B) \operatorname{copper sulphate} $ $ (C) \operatorname{copper nitrate} $ $ (D) \operatorname{copper nitrate} $

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30.	In the reaction PC	$Cl_3 + Cl_2 \rightarrow PCl_5$				
	(A) PCl <sub>3</sub> is acting	as reductant				
	(B) Cl <sub>2</sub> is acting as					
		Cl <sub>2</sub> are acting as redu				
	(D) both PCl <sub>3</sub> and	Cl <sub>2</sub> are acting as oxid	ant.			
		PART-C	C:BIO	LOGY		
31.	The internal (cellu	lar) energy reserve in	autotroph	s is :		
	(A) Glycogen	(B) Protein	(C)	Starch	(D) Fatty acid.	
32.	Normal blood pres	ssure range is				
	(A) 120/80	(B) 100/60	(C)	140/90	(D) 90/50	
33.	Movement of food	I through the lower reg	gion of oes	ophagus is du	ue to	
	(A) Lubrication by	saliva	(B)	Peristalsis		
	(C) Gravitational p	ull	(D)	All of these.		
34.	Heart beat origina	tes from :-				
	(A) Left atrium		(B)	Right ventric	le	
	(C) Pacemaker		(D)	Cadiac musc	cles	
35.	The white matter i	in a bird's dropping is	:-			
	(A) Calcium carbo	onate	(B)	Calcium sulp	hate	
	(C) Uric acid		(D)	Urea		
		Space	for rough w	ork		

**41.** Match the labelled parts of the given figure with the correct option.



Α	В	C	D	E
(A) Ovary	Thalamus	Filament	Sepal	Anther
(B) Ovary	Thalamus	Style	Sepal	Anther
(C) Ovule	Sepal	Style	Thalamus	Filament
(D) Ovule	Sepal	Style	Thalamus	Stamen

- 42. Oral contraceptives prevent
  - (A) Fertilization

- (B) Ovulation
- (C) Entrance of sperms in vagina
- (D) All of these
- 43. An organism with two unlike genes for a trait is called
  - (A) Homozygous
- (B) Heterozygous
- (C) Wild variety
- (D) Dominant variety.
- 44. Mendel conducted his famous breeding experiments by working on
  - (A) Drosophila
- (B) Pisum sativum
- (C) Escherichia coli (D) all of these.
- 45. The crossing of a homozygous tall plant with a dwarf would yield plants in the ratio of
  - (A) Two tall and two dwarf
  - (B) One homozygous tall, one homozygous dwarf and two heterozygous tall
  - (C) All homozygous dwarf
  - (D) All heterozygous tall.



## **PART-D: MENTAL ABILITY**

### Find the missing number:

97 78 43 67 46.

- 58 35 13 39
  - (A) 84
- (B) 80

- (C) 54
- (D) 48

- (B) 9

- (C) 6
- (D) 8

48.

- (A) 9
- (B) 8

(C) 6

- (D) 5
- 49. If in a code language MENTAL is coded as 417253, then how is TEN & ANT coded in that language?
  - (A) 572,271
- (B) 217,527
- (C) 572, 217
- (D) 217, 572
- **50.** If **REASON** is coded as 5 and **BELIEVED** as 7, what is the code number for **GOVERNMENT**?
  - (A) 6
- (B) 8

(C) 9

(D) 10

- 51. If E = 5 & SAFE = 31, then PINK = ?
  - (A) 41
- (B) 40

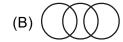
- (C) 50
- (D) 65

SAMP	LE PAPER (Medical Groun	d Zero)			[ 11 ]
52.	walks 30 metre. The	n moves 25 metre to his	right	He then turns to I	then turns to his left and his right again and walks direction is he now from
	(A) South-West	(B) South	(C)	North-West	(D) South-East
53.					owards South. Then, he reference to his starting
	(A) 5 km, North	(B) 5 km, North-East	(C)	7 km, East	(D) 7 km, West
54.	•	•		_	alked another two paces nich direction was he last
	(A) East	(B) North	(C)	South	(D) None of these
55.		graph, a man says to his ". How is the girl in the pl		•	nd-daughter of the elder e man ?
	(A) Niece	(B) Sister	(C)	Aunt	(D) Sister-in-law
56.		the photograph. Amar sa w is the girl's mother rel			ner is the only son of my
	(A) Mother	(B) Sister	(C)	Sister-in-law	(D) Grandmother
57.	T is the son of P.S. to T?	is the son of Q. T is ma	ırried	to R. R is Q's dau	ughter. How is S related
	(A) Brother	(B) Uncle	(C)	Father-in-law	(D) Brother-in-law
		Space for ro	ugh w	ork	

DIRECTIONS: (58 to 60) Each of these questions given below contains three group of things. You are to choose from the following four numbered diagrams, a diagram that depicts the correct relationship among the three groups of thing in each question.

58. Moon, Earth, Universe







(D) (O)

59. India, Pakistan, Asia



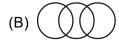






60. Batsman, Cricket, Stick









# ANSWER KEYS SAMPLE TEST PAPER

(For Students Going to Class 11<sup>™</sup> IN 2024)

**STREAM**: MEDICAL | COURSE OFFERED: MEDICAL GROUND ZERO

		-	HVCTCC	
_			PHYSICS	
1.	(A)	2. (A)	3. (D)	4. (B)
5.	(C)	6. (C)	7. (A)	8. (D)
9.	(D)	10. (B)	11. (D)	12. (C)
13.	(A)	14. (A)	15. (B)	
		CH	IEMISTRY	
16.	(C)	17. (D)	18. (D)	19. (C)
20.	(D)	21. (A)	22. (A)	23. (B)
24.	(A)	25. (B)	26. (C)	27. (B)
28.	(C)	29. (A)	30. (A)	
		_	TOLOGY	
	(2)		BIOLOGY	-
31.	(C)	32. (A)	33. (B)	34. (C)
35.	(C)	36. (B)	37. (D)	38. (C)
39.	(C)	40. (B)	41. (B)	42. (B)
43.	(B)	44. (B)	45. (D)	
		MEN	TAL ABILITY	
46.	(B)	47. (C)	48. (D)	49. (D)
<b>50</b> .	(C)	51. (C)	52. (D)	53. (B)
54.	(A)	55. (A)	56. (A)	57. (D)
<b>58</b> .	(A)	59. (A)	60. (D)	